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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is

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transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity.

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Answers Activity A. Student Exploration
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Student Exploration: Calorimetry Lab
Vocabulary: calorie, calorimeter, joule,
specific heat capacity Prior Knowledge
Questions (Do these BEFORE using the
Gizmo.) 1. The Latin word calor means

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“heat,” and meter comes from the Greek word meaning “to measure.” What do you think a calorimeter does? Measures heat 2. Where have you heard the word calorie before? ? What do you think a calorie

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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