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To say that a solution has a concentration of 5 M means that in 1 L of solution (not solvent) there would be 5 mol of solute: to prepare such a solution one would place 5 mol of NaCl in a 1-L flask, and then add whatever amount of water is necessary so that the

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composition and appearance throughout. Examples of homogeneous mixtures in the real world include rubbing alcohol (70% isopropyl alcohol, 30% water) and gasoline (a mixture of ...

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Question 2.

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Example 15.3.2. A 1.00 mol sample of NOCl was placed in a 2.00 L reactor and heated to 227°C until the system reached equilibrium. The contents of the reactor were then analyzed and found to contain 0.056 mol of Cl<sub>2</sub>. Calculate K at this temperature. The equation for the decomposition of NOCl to NO and Cl<sub>2</sub> is as follows:

$$2\text{NOCl} \rightleftharpoons 2\text{NO} + \text{Cl}_2$$

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