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Bonding and Molecular Structure 8-3 There are two types of repulsive forces between the two atoms. First, the nuclei repel because they are both positively charged. Second, the electrons repel because they are both negatively charged.

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8.1 The Covalent Bond

pages 240–247

Practice Problems page

244 Draw the Lewis

structure for each

molecule. 1. PH<sub>3</sub> H<sub>2</sub> H<sub>2</sub>

H— H H P respectively,

for single, double, and

triple P — — 2. H<sub>2</sub> S H

H H — H S S ...

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Chapter 8 Chemistry

Covalent Bonding

Chapter 8: Covalent

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Covalent Bonding

Single Covalent Bonds

When only one pair of

electrons is shared,

such as in a hydrogen

molecule, it is a single

covalent bond The

shared electron pair is

often referred to as the

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structures, Since we  
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8.2 The Nature of  
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however, indicates ...

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