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Second Semester - Notre dame Chemistry

Chemistry Exam 2nd Semester. combination reaction. decomposition reaction. combustion reaction. single replacement reaction. a chemical change in which two or more substances react to form.... a reaction in which a single compound breaks down to form two.... a chemical reaction that occurs when a substance reacts with o....

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This exam includes topics typically covered in the second semester of the two-semester general chemistry course. Topics include: Physical Properties and Intermolecular Forces. Phases and Phase Changes. Kinetics. Equilibrium. Acid/Base Chemistry (including pH and buffers) Thermodynamics. Electrochemistry.

2017 General Chemistry Exam - Second Term | ACS Exams

Chemistry - 2nd Semester Practice Test Thermochemistry (Ch.11): Multiple Choice 24) What is the amount of heat required to raise the temperature of 200.0 g of aluminum by 10°C? cal (specific heat of aluminum = 0.21 J/g°C) 24) A) 4200 cal B) 420 cal C) 42 000 cal D) 420 000 cal E) none of the above

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YHS Chemistry Second Semester Final Exam Study Guide 1. What mass of hydrogen gas can be produced by reacting 6.92 g of sodium in excess water? The reaction is: 2. Calculate the masses of silver chloride and dihydrogen monosulfide formed when 85.6 g of silver sulfide reacts with excess hydrochloric acid, HCl. (double replacement reaction) a. AgCl = 99.0g b. H₂S = 11.8g 3.

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These guides are closely aligned with chapters 12-17 and 21 and were designed for use in the second semester sequence of general chemistry. They can be used in a flipped-style classroom where students complete them before the lecture. Or they can be used to reinforce important topics learned in class.

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