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The quantities to consider in a chemical  
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12.1 Gases 1. motion 2. a. small b.  
forces c. random d. elastic; kinetic 3.  $KE = \frac{1}{2}mv^2$  4. Temperature 5. true 6. true  
7. false 8. true 9. true 10. false 11. true

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12. false 13. a 14. a 15. d 16. d 17. b 18.  
b ...

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your textbook, read about phase  
changes. Date Class STUDY GUIDE  
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and final phases for each phase change  
and making a check (V) in the correct  
energy column. Phase Change Phase  
Energy released required initial final SO  
Column B boiling point freezing point  
melting point

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1. Chapter 12 - Chemical Kinetics. 12.1  
Reaction Rates. A. Chemical kinetics 1.  
Study of the speed with which reactants  
are converted to products B. Reaction  
Rate 1. The change in concentration of a  
reactant or product per unit of time.  $t$  A  $t$

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t concentration of A at time t  
concentration of A at time t Rate. 2 1 2  
1.

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