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## **Collision Theory Gizmo Answer**

The Collision Theory Gizmo™ allows you to experiment with several factors that affect the rate at which

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reactants are transformed into products in a chemical reaction. You will need blue, green, and orange markers or colored pencils for the first part of this activity.

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Observe a chemical reaction with and without a catalyst. Determine the effects of concentration, temperature, surface area, and catalysts on reaction rates.

Reactant and product concentrations through time are recorded, and the speed of the simulation can be adjusted by the user.

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and product  
concentrations through  
time are recorded and  
the speed of the  
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adjusted by the user. I have noticed this year that many students skip these questions and it hurts their ability to connect to what the lesson is about.

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a direct instruction  
lesson, the  
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fantastic. It demonstrates the basics of collision theory, and then has experiments for students to run testing out the

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Enlightenment Rabies  
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collision theory. His  
boots looked so  
perfect. Two dark  
parabolas in a field of

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yellow, slight three-dimensional interest provided by the scurf strewn about Gizmo answer key collision theory.

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working on the concentration of a catalyst had the hardest task as they had to figure out what it was about the catalyst that was speeding the.

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Theory. NCVPS  
Chemistry Fall 2014  
Vocabulary: activated  
complex, catalyst,  
chemical reaction ...

## **Student Exploration- Collision Theory (ANSWER KEY) by ...**

Use collision theory in  
your answer. It will  
increase the surface  
area, making the  
particles more  
exposed, allowing  
them to react. Explain  
why all reactions have

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an activation energy. Collision theory states that there needs to be enough energy to break the bonds between molecules and smash the electron clouds together.

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Terms in this set (7)

More collisions increases the rate of reaction as the rate of a reaction depends on how often and how hard the reacting particles collide with each other.--

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several factors that  
affect the rate at which  
reactants  
are transformed into  
products in a chemical  
reaction. You will need  
blue, green, and  
orange markers or  
colored pencils for the  
first part of this activity

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