

## Measurements And Their Uncertainty Answers

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### Measurements And Their Uncertainty Answers Prentice Hall

Since 106.7 g has the most uncertainty ( $\pm 0.1$  g), the answer rounds off to one decimal place. The correct answer is 107.1 g and is read "one hundred and seven point one grams."

#### EXAMPLE EXERCISE 2.1 Uncertainty in Measurement

Circle the letter of the correct digit. In the measurement 43.52 cm, which digit is the most uncertain? a. 4 c. 5 b. 3 d. 2 16. Circle the letter of the correct number of significant figures in the measurement 6.80 m. a. 2 c. 4 b. 3 d. 5 17. List two situations in which measurements have an unlimited number of significant figures. 18.

### SECTION 3.1 MEASUREMENTS AND THEIR UNCERTAINTY

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#### Chemistry Measurements And Their Uncertainty Answer Key

Section 3.1 - Measurements and Their Uncertainty A measurement is a quantity that has both a number and a unit. The unit typically used in the sciences are those of the International System of Measurements (SI). In scientific notation, a given number is written as the product of two numbers: a coefficient and 10 raised to a power.

#### Chapter 3 - Scientific Measurement

3.1 Measurements and Their Uncertainty 3.2 The International System of Units 3.3 Conversion Problems 3.4 Density. Terms in this set (30) Measurement. A quantity that has both a number and a unit. Scientific Notation. A given number is written as the product of two numbers. Precision.

#### Chemistry Chapter 3 Scientific Measurement Flashcards ...

20. Give the Of significant figures in the following measurements, 3.85 x b. 17.30 cmS perform the operations and give in Standard form With the number of significant 37.2mL. 8.382 Ck. b. a.1Z9Y10 = 2.1

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Why Is there Uncertainty? • Measurements are performed with instruments, and no instrument can read to an infinite number of decimal places •Which of the balances shown has the greatest uncertainty in measurement? 11 Significant Figures in Measurements Significant figures in a measurement include all of the digits that are known, plus one

#### Chapter 3 Measurements and Their Scientific Uncertainty

Chapter 3 Guided Notes Name: \_\_\_\_ 3.1 Measurements and Their Uncertainty 1. A measurement is a quantity that has both a \_\_\_\_ and a \_\_\_\_ . 2. Scientific notation is a number written as the product of two numbers: a \_\_\_\_ and a \_\_\_\_ . The coefficient is also known as the 'first factor'.

#### Chapter 3 Guided Notes 3.1 Measurements and Their ...

As students complete their three measurements I take one student's paper (who has the measurements done correctly in terms of 1, 2, and 3 sig figs) and share on the overhead. I do this to stress that the first measurement should only have 1 number, the second 2 numbers, and the third 3 numbers.

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Knowing that the uncertainty in height measurement is  $\Delta h = 0.02$  m and the average human response time is  $\Delta t = 0.2$  s, estimate the uncertainty in your experimental  $g$ . Show equations and calculations.  $\Delta t = 0.2$  s so you take the farthest number away from the mean, 1.33 seconds and take  $0.2/1.33 = .15$  or 15% uncertainty.

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