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CHAPTER 3 REVIEW

Atoms: The Building
Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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CHAPTER 2 REVIEW
Measurements and
Calculations SECTION 1
SHORT ANSWER

Answer the following
questions in the space
provided. 1. Determine
whether each of the

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following is an example of observation and data, a theory, a hypothesis, a control, or a model.

observation and data
a. A research team records the rainfall in inches

2 Measurements and Calculations

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CHAPTER 10 REVIEW
States of Matter
SECTION 3 SHORT

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ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left.

b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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10 States of Matter -

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Chemistry Page

CHAPTER 1 REVIEW

Matter and Change

MIXED REVIEW SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Classify each of the

following as a

homogeneous or

heterogeneous

substance.

homogeneous a. sugar

homogeneous d.

plastic wrap

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homogeneous b. iron filings heterogeneous e. cement sidewalk heterogeneous c. granola bar

2. For each type of investigation, select the most appropriate ...

1 Matter and Change

- HUBBARD'S

CHEMISTRY

CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer the following questions in

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Answers

the space provided. 1.

c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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5 The Periodic Law

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