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Name Date Pd Unit 6 – Worksheet 4
Molecular Compounds Properties Basic
structural unit 1. Name each of the
following binary compounds of non-
metallic elements a. CBr 4 b. N₂ P₃ c.
PCl₃ d. ICl e. N₂ O f. SiF₄ g. GeH₄ h. N
2 Br 4 i. P₂ S₅ j. SeO₂ k. NH₃ l. SiO₂
2. Write the formula for the following
binary compounds of nonmetallic ...

U6 WS 4 Molecular Compounds (1).doc - Name Date Pd Unit 6 ...

Unit 6.4 Molecular Formulas and
Covalent Bonds Teacher: Dr. Van Der
Sluys a.k.a Dr. Bond (“Shaken not
stirred”) Objectives •Lewis dot
structures of –Molecular Compounds
–Bond Order –Polyatomic Ions •Cations
•Anions –Ionic Compounds Containing
Polyatomic Ions Molecular Compounds
•The formulas of molecular compounds

Unit 6.4 Molecular Formulas and Covalent Bonds

Unit 6: Molecular Compounds. Atoms.
Monatomic = 1 atom. Molecules.

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Molecule = 2 or more nonmetal . atoms acting as a . unit. Molecules. Diatomic = molecules of 2 of the same atoms. Molecular Compounds. Composed of molecules . Low melting and boiling points. Do not conduct electricity in H. 2 O Gases or liquids at room temp. 2 or more different ...

Unit 6: Molecular Compounds

Unit 6 Worksheet 4 Molecular Compounds Answers

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Online Library Unit 6 4 Molecular Compounds Answers Molecular compounds or covalent compounds are those in which the elements share electrons via covalent bonds. The only type of molecular compound a chemistry student is expected to be able to name is a binary covalent compound.

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6.2 Covalent Bonding and Molecular Compounds Flashcards ...

Figure 6.4.5 illustrates two of the molecular structures possible for hydrocarbons that have six carbon atoms. As you can see, compounds with the same molecular formula can have very different structures. The hybridization of the carbon atoms in these cases is sp^3 and the molecular geometry around each C atom is tetrahedral

Chapter 6.4: Naming Organic Compounds - Chemistry LibreTexts

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Name Date Pd Unit 6 - Worksheet 4.
Molecular Compounds. Properties. Basic structural unit 1. Name each of the following binary compounds of non-metallic elements

template

AFM image of 1,5,9-trioxo-13-azatriangulene and its chemical structure. A molecule is an electrically neutral group of two or more atoms held together by chemical bonds. Molecules are distinguished from ions by their lack of electrical charge.. In quantum physics, organic chemistry, and biochemistry, the distinction from ions is dropped and molecule is often used when referring to polyatomic ions.

Molecule - Wikipedia

Unit 6 Worksheet 4 Molecular Compounds File mole and formula unit worksheet.docx (docx) file mole problems worksheet.doc (doc) file percent composition, empirical, molecular formula worksheet.docx

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Unit 6 Worksheet 4 Molecular Compounds, Name ____ Class ...

For example, the ionic compound sodium oxalate is comprised of Na^+ and $\text{C}_2\text{O}_4^{2-}$ ions combined in a 2:1 ratio, and its formula is written as $\text{Na}_2\text{C}_2\text{O}_4$. The subscripts in this formula are not the smallest-possible whole numbers, as each can be divided by 2 to yield the empirical formula, NaCO_2 .

2.6 Molecular and Ionic Compounds - Chemistry

Unit 6 - Worksheet 4 Molecular
Compounds Properties Basic structural
unit 1. Name each of the following
binary compounds of non-metallic
elements a. CBr_4 b. N_2P_3 c. PCl_3 d. ICl e.
 N_2O f. SiF_4 g. GeH_4 h. N_2Br_4 i. P_2S_5 j.
 SeO_2 k. NH_3 l. SiO_2 ©Modeling
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Unit 6 - Worksheet 4

The compound consists mostly of carbon
and hydrogen atoms, so it is organic. B

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As with all organic compounds, C and H are written first in the molecular formula. Adding appropriate subscripts gives the molecular formula $C_2H_6O_2$.

6.1: Molecules and Molecular Compounds - Chemistry LibreTexts

Unit 6 - Worksheet 5 - Molecular Compounds

1. Describe at least 3 properties of molecular compounds:
2. Do aqueous solutions of molecular compounds conduct electricity? Explain why or why not, supporting your explanation with a particle diagram.
3. What is the basic structural unit (simplest particle) of a molecular compound? a) Draw an ...

Name Date Pd Unit 6 Worksheet 5 Molecular Compounds

Unit 3:1 Naming Compounds What's it Made of? Metal and non-metal

- 2 non-metals
- 3 or more elements polyatomic compound covalent compound ionic compound

USE GREEK PREFIXES Put prefixes in front of element names to tell

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how many atoms are there. Don't use "mono" for first name, but always for second name. Li 2S Metal and non-metal— ionic

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